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## CYCLOADDITION OF 1-AZIRINES WITH NITRILE OXIDES. FORMATION OF CARBODIIMIDES.

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In the course of our work on the symmetry-allowed thermal  $[\pi^4 + \pi^2]$  cycloadditions of the rigid C=N bond of 1-azirines,<sup>1,2</sup> we examined a 1,3-dipolar cycloaddition to the three-orbital  $4\pi$  electron system of nitrile oxides.<sup>3,4</sup> We discovered that aromatic nitrile oxides react exothermically with 1-azirines to furnish carbodiimides in > 80% isolated yields.

Thus when 3-methyl-2-phenyl-1-azirine (I)<sup>1</sup> was treated with 2,4,6-trimethylbenzonitrile oxide<sup>5</sup> in anhydrous ether at 0<sup>o</sup> for 15 min, and the product carefully purified by preparative layer chromatography (controlled atmosphere development), the carbodiimide (II) was obtained as white plates, mp 39-41 (decomp.). Its infrared spectrum (Nujol) showed N=C=N absorption at 2120 cm<sup>-1</sup> and C=O absorption at 1675 cm<sup>-1</sup>. Its nmr spectrum (in dry CDCl<sub>3</sub>)<sup>8</sup> showed absorptions at ( $\delta$  values) 1.54 (d, J = 7.0 Hz, 3H), 2.20 (s, 6H), 2.29 (s, 3H), 4.93 (q, J = 7.0 Hz, 1H), 6.78 (s, br, 2H), and 7.22 to 8.05 (m, 5H).

The carbodiimide was highly hygroscopic, and hydrolysis to the urea (III) mp 196<sup>0</sup>, proceeded rapidly and quantitatively.

A possible mechanism for the formation of the carbodiimide (Fig. 1) assumes the initial formation of a cycloadduct which undergoes ring cleavage and a 1,2shift of the R group. The initially formed cycloadduct is capable of valence tautomerism.



The generality of this transformation was established by preparation of carbodiimides from 2,3-diphenyl-1-azirine and 2-phenyl-1-azirine.<sup>10</sup> Further investigation on the direct spectral observation of the initial cycloadducts is in progress.

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